

William Clegg\* and Mark R. J. Elsegood†

Department of Chemistry, University of  
Newcastle upon Tyne, Newcastle upon Tyne  
NE1 7RU, England† Current address: Chemistry Department,  
Loughborough University, Loughborough  
Leicestershire LE11 3TU, England

Correspondence e-mail: w.clegg@ncl.ac.uk

## Key indicators

Single-crystal X-ray study  
 $T = 160$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.027  
 $wR$  factor = 0.068  
Data-to-parameter ratio = 10.7For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.(4*S*,5*S*)-4-Methyl-5-(4-tolylthiocarbonyl)-  
oxazolidin-2-one

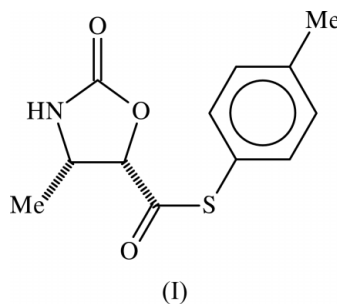
The title compound,  $\text{C}_{12}\text{H}_{13}\text{NO}_3\text{S}$ , is the product of a stereoselective epoxidation reaction, followed by cyclization and hydrolysis of the resulting epoxide. The heterocyclic ring is slightly twisted. Molecules are linked into a chain through intermolecular  $\text{N}-\text{H} \cdots \text{O}=\text{C}$  hydrogen-bonding interactions.

Received 10 June 2001  
Accepted 17 June 2002  
Online 21 June 2002

## Comment

The title compound, (I), was synthesized as part of a study of the stereocontrolled synthesis of *anti*- $\alpha$ -hydroxy- $\beta$ -amino acid derivatives (Ambroise *et al.*, 2002). Epoxidation of 1-arylthio-1-nitroalkenes was expected to give the corresponding epoxides, but instead oxazolidinones, such as (I), were isolated, presumably because of cyclization and hydrolysis during the work-up of the reaction products.

The crystal structure of (I) was determined in order to confirm the relative stereochemistry of the substituents on the heterocyclic ring; this was found to be *cis*. The product of the reaction, like the starting material, is enantiomerically pure. This is also indicated by NMR spectroscopy, and demonstrates that the reaction is highly stereoselective.

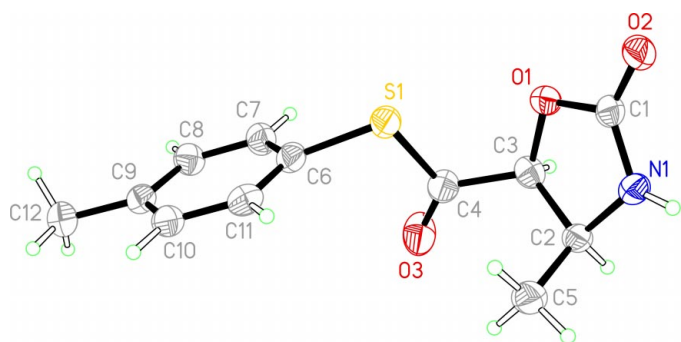


The bond lengths and angles of the heterocyclic ring are typical of those found in 39 related compounds in the April 2002 release of the Cambridge Structural Database (Allen & Kennard, 1993). The ring has a slight twist, as indicated by the torsion angles in Table 1; C2 lies 0.232 Å on one side and C3 0.040 Å on the other side of the plane defined by the N—C—O segment of the ring. This gives a conformation intermediate between twist and envelope (with C2 as the flap angle). The molecular structure is shown in Fig. 1.

An intermolecular hydrogen bond (Table 2) between the N—H and carbonyl groups links molecules into a chain parallel to the *b* axis.

## Experimental

The synthesis of the title compound is described by Ambroise *et al.* (2002).



**Figure 1**  
The molecular structure of (I), with atom labels and 50% probability ellipsoids for non-H atoms.

#### Crystal data

$C_{12}H_{13}NO_3S$   
 $M_r = 251.29$   
 Orthorhombic,  $P2_12_12_1$   
 $a = 5.5614$  (7) Å  
 $b = 7.8100$  (10) Å  
 $c = 27.520$  (5) Å  
 $V = 1195.3$  (3) Å<sup>3</sup>  
 $Z = 4$   
 $D_x = 1.396$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation  
 Cell parameters from 297 reflections  
 $\theta = 1.0$ – $23.3^\circ$   
 $\mu = 0.27$  mm<sup>-1</sup>  
 $T = 160$  (2) K  
 Block, colourless  
 $0.80 \times 0.64 \times 0.32$  mm

#### Data collection

Siemens SMART 1K CCD diffractometer  
 $\omega$  rotation with narrow frames  
 Absorption correction: multi-scan (XPREP in SHELXTL; Sheldrick, 1994)  
 $T_{\min} = 0.816$ ,  $T_{\max} = 0.891$   
 4745 measured reflections

1718 independent reflections  
 1714 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.036$   
 $\theta_{\text{max}} = 23.2^\circ$   
 $h = -6 \rightarrow 5$   
 $k = -8 \rightarrow 8$   
 $l = -30 \rightarrow 28$

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.027$   
 $wR(F^2) = 0.068$   
 $S = 1.11$   
 1718 reflections  
 160 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0296P)^2 + 0.3596P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.13$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.16$  e Å<sup>-3</sup>  
 Extinction correction: SHELXTL  
 Extinction coefficient: 0.019 (2)  
 Absolute structure: Flack (1983), 669 Friedel pairs  
 Flack parameter =  $-0.02$  (8)

**Table 1**

Selected geometric parameters (Å, °).

|             |             |             |             |
|-------------|-------------|-------------|-------------|
| O1—C1       | 1.362 (2)   | C1—O2       | 1.214 (2)   |
| O1—C3       | 1.445 (2)   | N1—C2       | 1.461 (3)   |
| C1—N1       | 1.330 (3)   | C2—C3       | 1.549 (3)   |
| C1—O1—C3    | 109.51 (15) | C1—N1—C2    | 113.50 (17) |
| O1—C1—N1    | 109.61 (17) | N1—C2—C3    | 99.26 (15)  |
| O1—C1—O2    | 120.73 (17) | O1—C3—C2    | 105.30 (15) |
| N1—C1—O2    | 129.67 (19) |             |             |
| C3—O1—C1—N1 | −1.7 (2)    | C1—O1—C3—C2 | 11.49 (19)  |
| O1—C1—N1—C2 | −10.0 (2)   | N1—C2—C3—O1 | −15.57 (18) |
| C1—N1—C2—C3 | 15.9 (2)    |             |             |

**Table 2**

Hydrogen-bonding geometry (Å, °).

| $D-H \cdots A$      | $D-H$    | $H \cdots A$ | $D \cdots A$ | $D-H \cdots A$ |
|---------------------|----------|--------------|--------------|----------------|
| $N1-H1 \cdots O2^i$ | 0.81 (2) | 2.07 (2)     | 2.859 (2)    | 163 (2)        |

Symmetry code: (i)  $-x, y - \frac{1}{2}, \frac{1}{2} - z$ .

H atoms attached to C atoms were placed geometrically and refined with a riding model (including free rotation about C—methyl bonds), and with  $U_{\text{iso}}$  constrained to be 1.2 (1.5 for methyl groups) times  $U_{\text{eq}}$  of the carrier atom. The positional parameters of the H atom on the oxazolidine N atom were refined freely. The relatively low  $\theta_{\text{max}}$  value is due to a short crystal-to-detector distance for this data collection, which was carried out before parameters were optimized for one of the first commercial CCD detector systems.

Data collection: SMART (Siemens, 1995); cell refinement: SMART; data reduction: SAINT (Siemens, 1995); program(s) used to solve structure: SHELXTL (Sheldrick, 1994); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and local programs.

The authors thank the EPSRC for financial support and Professor Richard Jackson for supplying the sample.

#### References

- Allen, F. H. & Kennard, O. (1993). *Chem. Des. Autom. News*, **8**, 1, 31–37.  
 Ambrose, L., Dumez, E., Szeki, A. & Jackson, R. F. W. (2002). *Synthesis*. In the press.  
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.  
 Sheldrick, G. M. (1994). *SHELXTL*. Version 5. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.  
 Siemens (1995). *SMART* and *SAINTE*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.